## TEXT SEARCH

=> d his 161

L61		PLUS' ENTERED AT 13:06 S L57 AND (L59 OR L60		
-> 4	que 161			
_> u	-	SEA FILE=HCAPLUS SPE= PN	ON ABB=ON PLU=ON	US20090108232/
L3	1	SEA FILE=REGISTRY SPE	C=ON ABB=ON PLU=ON	12031-66-2/RN
L4	1	SEA FILE=REGISTRY SPE	C=ON ABB=ON PLU=ON	1314-61-0/RN
L5	1	SEA FILE=REGISTRY SPE	C=ON ABB=ON PLU=ON	12057-24-8/RN
L6 L7	4831	SEA FILE=HCAPLUS SPE= SEL PLU=ON L3 1- NA		L3
L8	4918	SEA FILE=HCAPLUS SPE=		L7
L10		SEA FILE=HCAPLUS SPE=		L4
L11	13077	SEL PLU=ON L4 1- NA		11
L12	30017	SEA FILE=HCAPLUS SPE=		L11
L14	1824/	SEA FILE=HCAPLUS SPE=		L5
L15	00000	SEL PLU=ON L5 1- NA		- 1 F
L16		SEA FILE=HCAPLUS SPE=		L15
L17		SEA FILE=HCAPLUS SPE=		L14 AND L16
L19		SEA FILE=HCAPLUS SPE=		L6 AND L8
L20	18206	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	L10 AND L12
L21	39	SEA FILE=HCAPLUS SPE= AND L20	ON ABB=ON PLU=ON	L17 AND L19
L22			PLU=ON (MOLAR OR M	OLE)(4A)RATIO
L23	3	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	L21 AND L22
L24	1	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	L1 AND L23
L25	3	SEA FILE=HCAPLUS SPE= 05)	ON ABB=ON PLU=ON	(LIO2)(2W)(TA2
L26	2	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	0.975(3W)0.982
L27	18	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	GTOREQ(4A)0.97
L28	10	SEA FILE=HCAPLUS SPE= 2	ON ABB=ON PLU=ON	LTOREQ(4A)0.98
L29	0	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	L27 AND L28
L30	129	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	((LITHIUM OR
		DILITHIUM) (A) (OXIDE C	OR DIOXIDE) OR LIO2	OR O2LI OR
		LI20 OR OLI2) (5W) (TAN	ITALUM(A)OXIDE OR TA	205 OR O5TA)
L36	89176	SEA FILE=HCAPLUS SPE= MAX/CT	ON ABB=ON PLU=ON	BIREFRINGENCE+
L38	37641	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	BIREFRING?
L40	19976	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	0.0005
L41		SEA FILE=HCAPLUS SPE=		
L42		SEA FILE=HCAPLUS SPE=		
L43		SEA FILE=HCAPLUS SPE=L38)		L40(L)(L36 OR
L44	45	SEA FILE=HCAPLUS SPE= L38)	ON ABB=ON PLU=ON	L41(L)(L36 OR
L45	45	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	L43 AND L44
L46		SEA FILE=HCAPLUS SPE=		
	1	L38)		. , ,
L47	1 7		PLU=ON LENS? OR OP	
L48		SEA FILE=HCAPLUS SPE=		L21 AND L47
L49		SEA FILE=HCAPLUS SPE= OR L38)		L48 AND (L36
L50	1	SEA FILE=HCAPLUS SPE=		L49 AND ((L40
		OR L41 OR L42 OR L43		
L51		SEA FILE=HCAPLUS SPE=		
L52	5	SEA FILE=HCAPLUS SPE=	ON ABB=ON PLU=ON	L48 AND (L22

```
OR L36 OR L38)
L53
             3 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L52 AND ((L23
               OR L24 OR L25 OR L26 OR L27 OR L28 OR L29))
L54
             4 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L52 AND L30
             4 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L53 OR L54
L55
             1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L52 AND ((L40
L56
               OR L41 OR L42 OR L43 OR L44 OR L45 OR L46))
1.57
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               OR L50 OR L51 OR L52 OR L53 OR L54 OR L55 OR L56)
               OUE SPE=ON ABB=ON PLU=ON PY=<2005 NOT P/DT
L59
               QUE SPE=ON ABB=ON PLU=ON (PY=<2005 OR PRY=<2005 OR
L60
               AY = <2005 OR MY = <2005 OR REVIEW/DT) AND P/DT
L61
            14 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L57 AND (L59
               OR L60)
=> d his 169
     (FILE 'MEDLINE, BIOSIS, EMBASE' ENTERED AT 13:49:04 ON 07 MAY
     2010)
L69
             0 S L66 AND L67 AND L68
=> d que 169
             1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 12031-66-2/RN
L3
L4
             1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 1314-61-0/RN
            1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 12057-24-8/RN
L5
L7
               SEL PLU=ON L3 1- NAME :
                                              5 TERMS
L11
               SEL PLU=ON L4 1- NAME :
                                             12 TERMS
               SEL PLU=ON L5 1- NAME:
L15
                                             5 TERMS
L66
           145 SEA L7
L67
           262 SEA L11
           183 SEA L15
L68
             0 SEA L66 AND L67 AND L68
T.69
=> d his 187
     (FILE 'DISSABS, PASCAL, CONFSCI, JAPIO, WPIX' ENTERED AT 14:00:40
     ON 07 MAY 2010)
1.87
             3 S L85 OR L86
               SAV TEMP L87 PEE942MULT/A
    FILE 'HCAPLUS' ENTERED AT 14:08:48 ON 07 MAY 2010
               SAV TEMP L61 PEE942HCP/A
    FILE 'STNGUIDE' ENTERED AT 14:09:18 ON 07 MAY 2010
=> d que 187
             1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 12031-66-2/RN
L3
L4
             1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 1314-61-0/RN
T.5
             1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 12057-24-8/RN
               SEL PLU=ON L3 1- NAME :
1.7
                                              5 TERMS
               SEL PLU=ON L4 1- NAME :
                                             12 TERMS
T.11
L15
               SEL PLU=ON L5 1- NAME :
                                             5 TERMS
L22
               QUE SPE=ON ABB=ON PLU=ON (MOLAR OR MOLE) (4A) RATIO
L25
             3 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (LIO2)(2W)(TA2
L26
            2 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON 0.975(3W)0.982
L2.7
            18 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON GTOREQ(4A)0.97
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L28
            10 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON LTOREQ(4A)0.98
               2
            129 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON ((LITHIUM OR
L30
               DILITHIUM) (A) (OXIDE OR DIOXIDE) OR LIO2 OR O2LI OR
               LI2O OR OLI2) (5W) (TANTALUM(A) OXIDE OR TA2O5 OR O5TA)
         37641 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON BIREFRING?
L38
         19976 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON .-.0.0005
L40
         19976 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON 0.0005
T.41
               QUE SPE=ON ABB=ON PLU=ON PY=<2005 NOT P/DT
L59
               OUE SPE=ON ABB=ON PLU=ON (PY=<2005 OR PRY=<2005 OR
L60
               AY = <2005 OR MY = <2005 OR REVIEW/DT) AND P/DT
L70
          2760 SEA L7
         12031 SEA L11
L71
L72
          9963 SEA L15
           25 SEA L70 AND L71 AND L72
L73
L74
             5 SEA L73 AND L22
L75
             1 SEA L73 AND L38
L76
             1 SEA L74 AND L75
L77
             5 SEA (L74 OR L75 OR L76)
            1 SEA L77 AND ((L25 OR L26 OR L27 OR L28))
L78
L79
            2 SEA L77 AND L30
L80
            1 SEA L77 AND (L40 OR L41)
L81
           18 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON 0.0005(3A)(PLU
              S OR MINUS)
L83
            1 SEA L77 AND L81
            5 SEA (L77 OR L78 OR L79 OR L80) OR L83
            0 SEA L84 AND L59
L85
L86
            3 SEA L84 AND L60
L87
            3 SEA L85 OR L86
=> dup rem 161 169 187
L69 HAS NO ANSWERS
FILE 'HCAPLUS' ENTERED AT 14:12:26 ON 07 MAY 2010
USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.
PLEASE SEE "HELP USAGETERMS" FOR DETAILS.
COPYRIGHT (C) 2010 AMERICAN CHEMICAL SOCIETY (ACS)
FILE 'WPIX' ENTERED AT 14:12:26 ON 07 MAY 2010
COPYRIGHT (C) 2010 THOMSON REUTERS
PROCESSING COMPLETED FOR L61
PROCESSING COMPLETED FOR L69
PROCESSING COMPLETED FOR L87
            16 DUP REM L61 L69 L87 (1 DUPLICATE REMOVED)
               ANSWERS '1-14' FROM FILE HCAPLUS
               ANSWERS '15-16' FROM FILE WPIX
```

#### TEXT SEARCH RESULTS

=> d 188 1-16 ibib ed abs hitstr hitind

L88 ANSWER 1 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN DUPLICATE 1

ACCESSION NUMBER: 2005:472377 HCAPLUS Full-text

DOCUMENT NUMBER: 143:16264

TITLE: Optical material,

> optoelectronic part and optoelectronic appliance

INVENTOR(S): Kumatoriya, Makoto; Chiku, Shinichiro; Geho,

Mikio; Fujii, Takashi; Kitamura, Kenji;

Takekawa, Shunji; Nakamura, Masaru

National Institute for Materials Science, PATENT ASSIGNEE(S):

Japan

PCT Int. Appl., 20 pp. SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT	NO.	KIND	DATE	APPLICATION NO. DAT	ſΕ
WO 2005	_ 049897	A1	20050602	WO 2004-JP15046	
₩:	CA, CH, CN, ES, FI, GB, KE, KG, KP,	CO, CR, GD, GE, KR, KZ,	, CU, CZ, , GH, GM, , LC, LK,	SA, BB, BG, BR, BW, BY, BZ, DE, DK, DM, DZ, EC, EE, EG, HR, HU, ID, IL, IN, IS, JP, LR, LS, LT, LU, LV, MA, MD, NI, NO, NZ, OM, PG, PH, PL,	
R₩:	PT, RO, RU, TT, TZ, UA, BW, GH, GM, ZW, AM, AZ, CY, CZ, DE, MC, NL, PL,	SC, SD, UG, US, KE, LS, BY, KG, DK, EE, PT, RO,	, SE, SG, , UZ, VC, , MW, MZ, , KZ, MD, , ES, FI, , SE, SI,	SK, SL, SY, TJ, TM, TN, TR, VN, YU, ZA, ZM, ZW NA, SD, SL, SZ, TZ, UG, ZM, RU, TJ, TM, AT, BE, BG, CH, FR, GB, GR, HU, IE, IT, LU, SK, TR, BF, BJ, CF, CG, CI, NE, SN, TD, TG	
EP 1693	488			EP 2004-792290 200 101	
	DE, FR, GB 0108232	A1	20090430	0 US 2006-595942 200 052	
PRIORITY APP	LN. INFO.:			JP 2003-392870 A 200 112	
ASSIGNMENT H	ISTORY FOR U	S PATEN'	T AWATIARI	WO 2004-JP15046 W 200 103  BLE IN LSUS DISPLAY FORMAT	

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

Entered STN: 03 Jun 2005

The invention relates to an optical material that is not affected by environmental changes, exhibiting a birefringence falling within the range of .+-.0. 0005; and an optoelectronic part and optoelectronic appliance including the optical material. There is provided an optical material of lithium tantalate characterized in that in the lithium tantalate the molar composition ratio of lithium oxide to

tantalum oxide (LiO2/Ta2O5) is in the range of 0.975 to 0. 982. Since an optical material of high refractive index can be used in an aptical system, the lens thickness can be reduced at an unchanged focal length. As a result, by the use of 1808 with such characteristics, not only can optomic parts having realized higher levels of compactness, thin model and function enhancement be provided but also optoelectronic appliances having these characteristics can be presented.

12031-66-2, Lithium tantalate TТ RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses) (optical material and optoelectronic application) RN 12031-66-2 HCAPLUS

CNLithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Со	mponent   	Ratio	   Re	Component egistry Number
	======+:		====+====	
0	1	3	1	17778-80-2
Ta		1	1	7440-25-7
Li		1		7439-93-2
IT	•	Tantalum oxide	roaction	ng.

12057-24-8, Lithium oxide, reactions

RL: RCT (Reactant); RACT (Reactant or reagent) (optical material and optoelectronic application)

RN 1314-61-0 HCAPLUS

Tantalum oxide (Ta2O5) (CA INDEX NAME) CN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 12057-24-8 HCAPLUS

CN Lithium oxide (Li2O) (CA INDEX NAME)

Li\_ O\_ Li

```
ICM C30B029-30
T.C.
     ICS G02B001-00
CC
     73-12 (Optical, Electron, and Mass Spectroscopy and Other Related
    Properties)
ST
    optical material optoelectronic lens
    Lenses
       Optical materials
        (optical material and optoelectronic
        application)
TТ
     12031-66-2, Lithium tantalate
     RL: PRP (Properties); TEM (Technical or engineered material use);
     USES (Uses)
        (optical material and optoelectronic
        application)
     1314-61-0, Tantalum oxide
ΙT
     12057-24-8, Lithium oxide, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (optical material and optoelectronic
        application)
REFERENCE COUNT:
                         8
                               THERE ARE 8 CITED REFERENCES AVAILABLE
                               FOR THIS RECORD. ALL CITATIONS AVAILABLE
                               IN THE RE FORMAT
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L88 ANSWER 2 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN ACCESSION NUMBER: 2005:1172105 HCAPLUS Full-text DOCUMENT NUMBER: 143:413341

TITLE: Method and apparatus for manufacture of

optical devices

INVENTOR(S): Okamoto, Tsutomu PATENT ASSIGNEE(S):

Sony Corp., Japan Jpn. Kokai Tokkyo Koho, 10 pp. SOURCE:

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE . Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2005309161	A	20051104	JP 2004-127182	
				2004
				0422
			<	
PRIORITY APPLN. INFO.:			JP 2004-127182	
				2004
				0422
			<	

F.D Entered STN: 04 Nov 2005

Apparatus for manufacture of the devices includes a container for placing raw material AB powder, a support for placing the substrate over the container, a covering for the container and the support, and an outer container for high-temperature heat treatment of the covered materials. Manufacture of optical devices by vapor phase diffusion of oxides over substrates are claimed. Ferroelec. material may be deposited on the substrate by vapor transport equilibration.

1314-61-0, Tantalum oxide

12057-24-8, Lithium oxide, uses

RL: PEP (Physical, engineering or chemical process); PYP (Physical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(method and apparatus for manufacture of optical devices by vapor surface treatment of substrates)

1314-61-0 HCAPLUS RN

Tantalum oxide (Ta2O5) (CA INDEX NAME) CN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

12057-24-8 HCAPLUS

CN Lithium oxide (Li20) (CA INDEX NAME)

Li\_\_O\_\_Li

ΤТ 12031-66-2, Lithium tantalate

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(substrate; method and apparatus for manufacture of optical

devices by vapor surface treatment of substrates)

12031-66-2 HCAPLUS RN

Lithium tantalum oxide (LiTaO3) (CA INDEX NAME) CN

Component	    +	Ratio	    +-	Component Registry Number
0	 	3	 	17778-80-2
Ta	1	1		7440-25-7
Li	1	1		7439-93-2

ICM G02F001-37 IC

73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

ST substrate surface vapor treatment optical device; vapor transport equilibration ferroelec layer optical device substrate Ferroelectric materials ΙT (method and apparatus for manufacture of optical devices by vapor surface treatment of substrates) TT Optical instruments (nonlinear; method and apparatus for manufacture of optical devices by vapor surface treatment of substrates) ΙT Surface treatment (vapor transport equilibration; method and apparatus for manufacture of optical devices by vapor surface treatment of substrates) 1314-61-0, Tantalum oxide TТ 12057-24-8, Lithium oxide, uses RL: PEP (Physical, engineering or chemical process); PYP (Physical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses) (method and apparatus for manufacture of optical devices by vapor surface treatment of substrates) TТ 12031-66-2, Lithium tantalate RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses) (substrate; method and apparatus for manufacture of optical devices by vapor surface treatment of substrates) L88 ANSWER 3 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN ACCESSION NUMBER: 2005:975836 HCAPLUS Full-text DOCUMENT NUMBER: 143:276834 TITLE: Manufacture of defect-free single crystals by Czochralski method without tailing process Ito, Takeshi; Hanyu, Masayuki; Matsukura, INVENTOR(S): Makoto; Natori, Masaaki; Nakamura, Osamu; Furukawa, Yasunori; Matsumura, Sadao Oxide Corporation, Japan PATENT ASSIGNEE(S): Jpn. Kokai Tokkyo Koho, 14 pp. SOURCE: CODEN: JKXXAF DOCUMENT TYPE: Patent LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION: DATE PATENT NO. KIND DATE APPLICATION NO. \_\_\_\_\_ \_\_\_\_ -----JP 2005239442 A 20050908 JP 2004-47794 2004 0224 JP 2004-47794 PRIORITY APPLN. INFO.: 2004 0224 <--ED Entered STN: 08 Sep 2005 In preparation of single crystals with different composition between molten liqs. and crystals, after formation of straight body parts is finished, parts of the crystals being lifted (or stopped) are melted (by heating the molten liqs.) to be released from the liqs. The process is effective for manufacturing of twin- or microcrack-free ferroelec. crystals (e.g., Li tantalate, Li niobate, BaSr niobate) useful for optical communication devices. 1314-61-0, Tantalum oxide ( Ta2O5) 12O57-24-8, Lithium oxide (Li2O), processes RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)

(manufacture of defect-free oxide single crystals by Czochralski

method without tailing process)

```
RN 1314-61-0 HCAPLUS
CN Tantalum oxide (Ta205) (CA INDEX NAME)

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
RN 12057-24-8 HCAPLUS
CN Lithium oxide (Li20) (CA INDEX NAME)
```

Li\_\_O\_\_Li

#### IT 12031-66-2P, Lithium tantalate

RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); PYP (Physical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)

(manufacture of defect-free oxide single crystals by Czochralski method without tailing process)

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component		Ratio		Component
				Registry Number
	==+==		===+=	
0		3		17778-80-2
Ta		1		7440-25-7
Li		1		7439-93-2

IC ICM C30B015-22

ICS C30B029-30

CC 76-3 (Electric Phenomena)

Section cross-reference(s): 57

ST defect free ferroelec oxide crystal growth Czochralski;
lithium tantalate niobate crystal tailing free
Czochralski; melting lifting single crystal twin microcrack
prevention

IT 554-13-2, Lithium carbonate 1313-96-8, Niobium oxide (Nb2O5)

1314-61-0, Tantalum oxide (

Ta205) 12057-24-8, Lithium

oxide (Li20), processes 107251-85-4, Barium

niobium strontium oxide (Ba0.2NbSr0.303)

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)

(manufacture of defect-free oxide single crystals by Czochralski method without tailing process)

IT 12031-63-9P, Lithium niobate 12031-66-2P,

#### Lithium tantalate

RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); PYP (Physical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)

(manufacture of defect-free oxide single crystals by Czochralski method without tailing process)

L88 ANSWER 4 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN ACCESSION NUMBER: 2005:1283325 HCAPLUS Full-text

DOCUMENT NUMBER: 144:118343

TITLE: Method for preparing near-stoichiometric

lithium tantalate wafer

INVENTOR(S): Wang, Haili; Xia, Changtai; Xu, Jun; Hang,

Yin; Zhang, Lianhan; Liu, Junfang; Zhu,

Yueqin; He, Xiaoming

PATENT ASSIGNEE(S): Shanghai Institute of Optics and Fine

Mechanics, Chinese Academy of Sciences, Peop.

Rep. China

SOURCE: Faming Zhuanli Shenqing Gongkai Shuomingshu, 7

pp.

CODEN: CNXXEV

DOCUMENT TYPE: Patent
LANGUAGE: Chinese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1621577	A	20050601	CN 2004-10067130	
				2004 1013
			<	
PRIORITY APPLN. INFO.:			CN 2004-10067130	
				2004 1013
				1013

<--

ED Entered STN: 08 Dec 2005

The title method comprises: (1) placing mixed LiTaO3 and Li3TaO4 blocks having pores in a platinum crucible, (2) setting or hanging lithium tantalate wafer having the same composition on the platinum wire, (3) covering with a platinum sheet which is covered by mixed LiTaO3 and Li3TaO4 powder and thermocouple, (4) sealing with a platinum cap on the top of the crucible, and (5) placing the crucible in a resistance furnace, heating to 1000-1400ÅC, and maintaining the temperature for 1-200 h. By lithium ion diffusion, the crystal composition approaches the stoichiometric proportion. The method is simple, and the produced wafer has wide application in optical waveguide, photoelec. switch, periodic polarization, integrated photoelec. device and other fields.

IT 12031-66-2, Lithium tantalate

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)

(method for preparing near-stoichiometric lithium
tantalate wafer)

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component		Ratio		Component
				Registry Number
========	==+==	=========	===+=	
0		3		17778-80-2
Ta		1		7440-25-7
Li		1		7439-93-2

IT 1314-61-0, Tantalum pentoxide

RL: RCT (Reactant); RACT (Reactant or reagent)
(method for preparing near-stoichiometric lithium
tantalate wafer)

RN 1314-61-0 HCAPLUS

CN Tantalum oxide (Ta2O5) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 12057-24-8P, Lithium oxide,

preparation

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(method for preparing near-stoichiometric lithium
tantalate wafer)

RN 12057-24-8 HCAPLUS

CN Lithium oxide (Li20) (CA INDEX NAME)

```
TC
    ICM C30B029-30
    75-1 (Crystallography and Liquid Crystals)
    Section cross-reference(s): 73
    near stoichiometric lithium tantalate wafer
ST
    prepn
    Optical waveguides
ΤT
    Photoelectric devices
    Sintering
        (method for preparing near-stoichiometric lithium
       tantalate wafer)
    12031-90-2P, Lithium tentalum oxide (Li3TaO4)
IT
    RL: CPS (Chemical process); PEP (Physical, engineering or chemical
    process); SPN (Synthetic preparation); PREP (Preparation); PROC
     (Process)
        (method for preparing near-stoichiometric lithium
        tantalate wafer)
    12031-66-2, Lithium tantalate
TΤ
    RL: CPS (Chemical process); PEP (Physical, engineering or chemical
    process); TEM (Technical or engineered material use); PROC
     (Process); USES (Uses)
        (method for preparing near-stoichiometric lithium
       tantalate wafer)
    1308-38-9, Chromic oxide, uses 1309-37-1, Ferric oxide, uses
    1309-48-4, Magnesium oxide, uses 1312-43-2, Indium trioxide
    1313-97-9, Neodymium sesquioxide 1314-13-2, Zinc oxide, uses
    1314-37-0, Ytterbium oxide 1317-34-6, Manganic oxide
    1344-70-3, Copper oxide 1345-13-7, Cerous oxide
                                                       12060-08-1,
    Scandium oxide 12061-16-4, Erbium sesquioxide
    RL: MOA (Modifier or additive use); USES (Uses)
        (method for preparing near-stoichiometric lithium
       tantalate wafer)
    554-13-2, Lithium carbonate 1314-61-0,
TТ
    Tantalum pentoxide
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (method for preparing near-stoichiometric lithium
       tantalate wafer)
    12057-24-8P, Lithium oxide,
    preparation
    RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (method for preparing near-stoichiometric lithium
       tantalate wafer)
L88 ANSWER 5 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN
ACCESSION NUMBER: 2003:260910 HCAPLUS <u>Full-text</u>
DOCUMENT NUMBER:
                        138:278488
TITLE:
                        Pretreatment method for hologram recording
                        medium
                        Kitamura, Kenji; Takekawa, Shunji; Nakamura,
INVENTOR(S):
                        Masaru; Yamaji, Takashi; Hatano, Hideki
PATENT ASSIGNEE(S):
                        Independent Administrative Institution
                        National Institute for Materials Science,
                        Japan
SOURCE:
                        U.S. Pat. Appl. Publ., 10 pp.
                        CODEN: USXXCO
DOCUMENT TYPE:
                        Patent
LANGUAGE:
                        English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
    PATENT NO.
                      KIND DATE
                                         APPLICATION NO.
                                                                  DATE
                        ____
                               _____
                                           _____
```

US 2002-235853

US 20030064294 A1 20030403

2002 0906 JP 2003084652 Α 20030319 JP 2001-272499 2001 0907 <--JP 3728410 B2 20051221 PRIORITY APPLN. INFO.: JP 2001-272499 2001 0907

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

ED Entered STN: 04 Apr 2003

Disclosed is a pretreatment method for a hologram recording medium used in the hologram recording method in which information signals loaded on signal beam are recorded by injecting coherent signal beam and reference beam to the hologram recording medium which is exposed to first light having first wavelength of UV band or short-wavelength visible light band in advance in order to generate light-induced absorption, wherein the coherent signal beam and reference beam each having longer wavelength than the first wavelength. The pretreatment method comprises subjecting the hologram recording medium to oxidation treatment prior to the irradiation of the first light has been completed.

IT 1314-61-0, Tantalum oxide

12031-66-2, Lithium tantalate

12057-24-8, Lithium oxide, uses

RL: TEM (Technical or engineered material use); USES (Uses) (pretreatment method for hologram recording medium containing)

RN 1314-61-0 HCAPLUS

CN Tantalum oxide (Ta2O5) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component		Ratio		Component
				Registry Number
	==+==		===+==	
0	1	3	1	17778-80-2
Ta	1	1	1	7440-25-7
Li	1	1	1	7439-93-2

RN 12057-24-8 HCAPLUS

CN Lithium oxide (Li20) (CA INDEX NAME)

Li\_ O\_ Li

```
IC ICM G03H001-04
```

INCL 430001000; 430002000; 359007000; 430394000

CC 74-8 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

IT Optical reflectors

(pretreatment method for hologram recording medium containing)

IT 1313-96-8, Niobium oxide 1314-61-0, Tantalum oxide 12031-63-9, Lithium niobate (LiNbO3)

12031-66-2, Lithium tantalate

12057-24-8, Lithium oxide, uses

195144-63-9, Lithium oxide(LiO2)

RL: TEM (Technical or engineered material use); USES (Uses) (pretreatment method for hologram recording medium containing)

L88 ANSWER 6 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN

2002:439054 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 137:25970

TITLE: Lithium tantalate single

crystals and optical devices using

INVENTOR(S): Miyamoto, Akio; Kitamura, Kenji; Furukawa,

Yasunori; Takekawa, Shunji

PATENT ASSIGNEE(S): Hitachi Metals, Ltd., Japan; National

Institute for Research In Inorganic Materials

SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
 JP 2002167297	A	20020611	JP 2000-363531	
			<	2000 1129
PRIORITY APPLN. INFO.:			JP 2000-363531	2000 1129
			/	

Entered STN: 11 Jun 2002 ED

The single crystals are obtained from Li-excess melts and show Na content ≤10 ppm and AΒ molar ratio Li20/(Ta205 + Li20) 0.4900-0.5200. The crystals are useful for quasiphase-matched 2nd-harmonic generation (QPM-SHG) devices, gas detectors, etc. The crystals show decreased optical absorption at 280-320 nm and good resistance to optical damages at <390 nm.

1314-61-0, Tantalum oxide ( ΙT Ta2O5) 12031-66-2, Lithium

tantalum oxide (LiTaO3)

12057-24-8, Lithium oxide, uses

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(lithium tantalate single crystals with

decreased UV absorption for optical devices)

1314-61-0 HCAPLUS RN

CN Tantalum oxide (Ta2O5) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

12031-66-2 HCAPLUS

Lithium tantalum oxide (LiTaO3) (CA INDEX NAME) CN

Component	 	Ratio		Component Registry Number
	==+==		===+=	
0		3		17778-80-2
Ta		1		7440-25-7
Li		1	1	7439-93-2

12057-24-8 HCAPLUS RN

CN Lithium oxide (Li2O) (CA INDEX NAME)

Li\_\_O\_\_Li

ICM C30B029-30

ICS G02B001-02; G02F001-03; G02F001-355; G02F001-37

10/595,942-330158-EIC SEARCH CC73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties) lithium tantalate single crystal optical device; nonlinear optical device lithium tantalate single crystal; UV absorption decrease lithium tantalate single crystal Optical materials ΙT (lithium tantalate single crystals with decreased UV absorption for optical devices) ΙT Optical instruments (nonlinear; lithium tantalate single crystals with decreased UV absorption for optical devices) 7440-23-5, Sodium, miscellaneous ΤТ RL: MSC (Miscellaneous) (content-controlled; lithium tantalate single crystals with decreased UV absorption for optical devices) 1314-61-0, Tantalum oxide ( Ta2O5) 12O31-66-2, Lithium tantalum oxíde (LiTaO3) 12057-24-8, Lithium oxide, uses RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses) (lithium tantalate single crystals with decreased UV absorption for optical devices) OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS) L88 ANSWER 7 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN ACCESSION NUMBER: 2001:397790 HCAPLUS Full-text DOCUMENT NUMBER: 135:12383 TITLE: Single crystal of lithium niobate or tantalate and its optical element, and process and apparatus for producing an oxide single crystal Kitamura, Kenji; Furukawa, Yasunori; Takekawa, INVENTOR(S): Shunji; Kimura, Shigeyuki PATENT ASSIGNEE(S): National Institute for Research In Organic Materials, Japan SOURCE: U.S. Pat. Appl. Publ., 31 pp., Division of U.S. Ser. No. 521,899. CODEN: USXXCO DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: 2 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
 US 20010001944	A1	20010531	us 2001-754187	2001 0105
			<	
US 6464777	В2	20021015		
US 6673330	В1	20040106	US 2000-521899	
				2000 0309
			<	
JP 2001287999	A	20011016	JP 2000-341130	
				2000
				1108
			<	
JP 4107365	В2	20080625		
JP 2003267798	A	20030925	JP 2000-341132	
				2000

						1108
0005060606	_	00051010		<		
JP 2007269626	A	20071018	JP	2007-100442		2007
						0406
				<		
JP 2008176335	A	20080731	JP	2008-39835		
						2008
				<		0221
PRIORITY APPLN. INFO.:			.TP	1999-84999	А	
intentiti intim. into			01	1999 01999		1999
						0326
				<		
			JP	1999-317565	A	1000
						1999 1109
				<		1109
			JP	1999-317572	А	
						1999
						1109
			110	<	7.0	
			05	2000-521899	A3	2000
						0309
				<		
			JP	2000-341130	A3	
						2000
						1108
			TD	< 2000-341132	A3	
			UP	2000-341132	AJ	2000
						1108
				<		

ED Entered STN: 03 Jun 2001

A single crystal of Li niobate or tantalate is grown from a melt of a composition AB having an excessive Li over its stoichiometric composition, and having a molar fraction of Li20/(Nb205+Li20) or Li20/ (Ta205+Li20) within a range of at least 0.490 and <0.500, wherein at least one element selected from the group consisting of Mg, Zn, Sc and In is contained in an amount of from 0.1 to 3.0 mol based on the total amount of the at least one element, Nb and Li, or the total amount of the at least one element, Ta and Li. A process is described for producing an oxide single crystal by rotation pulling by a double crucible made of a noble metal consisting of an outer crucible made of a noble metal, and a cylindrical inner crucible for intersecting the surface of a melt in the outer crucible and connecting the melt at the bottom of the melt. The process comprises pulling a single crystal from the inner crucible while directly measuring the weight of the growing crystal for growing, simultaneously supplying a gas into a closed container, supplying a powder material preserved in the closed container between the outer crucible and the inner crucible through a supply tube in the same amount by weight as the crystal growth, and growing the crystal by rotating the double crucible. IΤ 12031-66-2, Lithium tantalate

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (single crystal of lithium niobate or tantalate and

optical element, and process and apparatus for producing oxide single crystal)

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component	 	Ratio	[ [	Component Registry Number
	==+==		===+=:	
0		3	- 1	17778-80-2
Ta		1	- 1	7440-25-7
Li		1		7439-93-2

IT 1314-61-0, Tantalum oxide (

```
12057-24-8, Lithium
     ta2o5)
     oxide (li2o), processes
     RL: PEP (Physical, engineering or chemical process); PROC
     (Process)
        (single crystal of lithium niobate or tantalate grown from melt
       of composition having molar fraction of Li20/(Nb205+Li20) or Li20/
        (Ta2O5+Li2O) within range of at least 0.490 and <0.500)
RN
    1314-61-0 HCAPLUS
    Tantalum oxide (Ta2O5) (CA INDEX NAME)
CN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
RN
   12057-24-8 HCAPLUS
CN
    Lithium oxide (Li2O) (CA INDEX NAME)
 Li_0_Li
    ICM C30B015-00
INCL 117013000
     75-1 (Crystallography and Liquid Crystals)
     Section cross-reference(s): 73
ST
     lithium niobate tantalate crystal optical element; oxide
     crystal growth app noble metal double crucible; polarization
     inversion optical element lithium niobate tantalate
    Czochralski crystal growth
TТ
    Czochralski crystal growth apparatus
     Electrooptical materials
     Nonlinear optical materials
        (single crystal of lithium niobate or tantalate and
       optical element, and process and apparatus for producing
       oxide single crystal)
ΤТ
    Oxides (inorganic), processes
     RL: PEP (Physical, engineering or chemical process); PROC
     (Process)
        (single crystal of lithium niobate or tantalate and
       optical element, and process and apparatus for producing
       oxide single crystal)
    12031-63-9, Lithium niobate 12031-66-2,
IΤ
     Lithium tantalate
     RL: DEV (Device component use); PEP (Physical, engineering or
     chemical process); PROC (Process); USES (Uses)
        (single crystal of lithium niobate or tantalate and
       optical element, and process and apparatus for producing
       oxide single crystal)
     1313-96-8, Niobium oxide (nb2o5) 1314-61-0,
IT
     Tantalum oxide (ta2o5)
     12057-24-8, Lithium oxide (
     li2o), processes
     RL: PEP (Physical, engineering or chemical process); PROC
     (Process)
        (single crystal of lithium niobate or tantalate grown from melt
        of composition having molar fraction of Li20/(Nb205+Li20) or Li20/
        (Ta205+Li20) within range of at least 0.490 and <0.500)
OS.CITING REF COUNT:
                             THERE ARE 9 CAPLUS RECORDS THAT CITE
                       9
                               THIS RECORD (12 CITINGS)
L88 ANSWER 8 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN
ACCESSION NUMBER:
                        2001:821445 HCAPLUS Full-text
DOCUMENT NUMBER:
                        136:76838
TITLE:
                        Nearly stoichiometric LiTaO3 for bulk
                        quasi-phase-matched devices
AUTHOR(S):
                        Furukawa, Y.; Nakamura, M.; Takekawa, S.;
                         Kitamura, K.; Hatanaka, T.; Nakamura, K.; Ito,
```

H.; Alexandrovski, A.; Fejer, M. M.

CORPORATE SOURCE: Oxide Corporation, Yamanashi, 408-0044, Japan

SOURCE: Trends in Optics and Photonics (2001

), 50(Advanced Solid-State Lasers), 685-687

CODEN: TOPRBS

PUBLISHER: Optical Society of America

DOCUMENT TYPE: Journal LANGUAGE: English ED Entered STN: 12 Nov 2001

AB Nearly stoichiometric LiTaO3 (SLT) crystals exhibited a high photorefractive damage resistance along with a negligibly small green-induced IR absorption, even without the addition of MgO dopants. A bulk periodically poled device was successfully fabricated using a 3 mm-thick SLT crystal.

IT 12031-66-2D, Lithium tantalum

oxide (LiTaO3), nearly stoichiometric

RL: DEV (Device component use); PRP (Properties); USES (Uses) (nearly stoichiometric LiTaO3 for bulk quasi-phase-matched devices)

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component	 	Ratio		Component Registry Number
	==+==:		===+=	
0	1	3		17778-80-2
Ta	1	1		7440-25-7
Li	1	1		7439-93-2

IT 1314-61-0, Tantalum oxide

12057-24-8, Lithium oxide, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation using; nearly stoichiometric LiTaO3 for bulk quasi-phase-matched devices)

RN 1314-61-0 HCAPLUS

CN Tantalum oxide (Ta2O5) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 12057-24-8 HCAPLUS

CN Lithium oxide (Li20) (CA INDEX NAME)

Li\_\_O\_\_Li

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CC 73-10 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
```

ST nearly stoichiometric lithium tantalate quasi

phase matched device

IT Optical instruments

(quasi-phase-matched; nearly stoichiometric LiTaO3 for bulk quasi-phase-matched devices)

IT 12031-66-2D, Lithium tantalum

oxide (LiTaO3), nearly stoichiometric

RL: DEV (Device component use); PRP (Properties); USES (Uses) (nearly stoichiometric LiTaO3 for bulk quasi-phase-matched devices)

IT 1314-61-0, Tantalum oxide

12057-24-8, Lithium oxide, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation using; nearly stoichiometric LiTaO3 for bulk quasi-phase-matched devices)

OS.CITING REF COUNT: 6 THERE ARE 6 CAPLUS RECORDS THAT CITE

THIS RECORD (6 CITINGS)

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE

FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L88 ANSWER 9 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN ACCESSION NUMBER: 2000:721528 HCAPLUS Full-text

DOCUMENT NUMBER: 134:48858

TITLE: Domain switching performance of stoichiometric

LiTaO3 for bulk quasi-phase matching devices Kitamura, Kenji; Furukawa, Yasunori; Takekawa,

Shunji; Hatanaka, Takaaki; Ito, Hiromasa;

Gopalan, Verkatraman

CORPORATE SOURCE: National Institute for Research in Inorganic

Materials, Tsukuba-shi, 305-0044, Japan

SOURCE: OSA Trends in Optics and Photonics Series (

2000), 34(Advanced Solid State

Lasers), 321-323

CODEN: OTOPFZ; ISSN: 1094-5695

PUBLISHER: Optical Society of America

DOCUMENT TYPE: Journal LANGUAGE: English ED Entered STN: 13 Oct 2000

The authors grew near stoichiometric LiTaO3 crystal from a Li-rich melt using a novel double crucible Czochralski method. The ferroelec. domain shape and domain wall smoothness were compared between the conventional and near stoichiometric LiTaO3 crystals. The domain shape under the elec. field at room temperature strongly depended on the densities of nonstoichiometric defects. The domain shape in the conventional LiTaO3 is basically triangular while it is hexagonal in the stoichiometric LiTaO3. The sides of hexagon in the stoichiometric LiTaO3 are perpendicular to the crystallog. X axes, i.e., parallel to the X faces. Therefore, considerably smooth domain walls can be obtain in the stoichiometric LiTaO3 when the periodical domain structure is designed as each domain elongates along the Y axis. This result is promising a great improvement by using stoichiometric LiTaO3 in fabricating quasi-phase matching wavelength conversion devices with high performances.

IT 12031-66-2, Lithium tantalate

LiTaO3

AUTHOR(S):

RL: DEV (Device component use); PRP (Properties); USES (Uses) (domain switching performance of stoichiometric LiTaO3 for bulk quasi-phase matching devices)

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component		Ratio	   R€	Component egistry Number
=========	==+===		===+====	==========
0		3	1	17778-80-2
Ta		1	[	7440-25-7
Li	1	1	1	7439-93-2

IT 1314-61-0, Tantalum pentoxide

12057-24-8, Lithium oxide, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(domain switching performance of stoichiometric LiTaO3 prepared using)

RN 1314-61-0 HCAPLUS

CN Tantalum oxide (Ta2O5) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 12057-24-8 HCAPLUS

CN Lithium oxide (Li20) (CA INDEX NAME)

Li\_O\_Li

CC73-10 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

domain switching phase matching stoichiometric lithium tantalate

Electric field effects TT

Ferroelectric domain

Interface roughness

Nonlinear optical properties

Optical harmonic generation

(domain switching performance of stoichiometric LiTaO3 for bulk quasi-phase matching devices)

ТТ 12031-66-2, Lithium tantalate

LiTa03

RL: DEV (Device component use); PRP (Properties); USES (Uses) (domain switching performance of stoichiometric LiTaO3 for bulk quasi-phase matching devices)

554-13-2, Lithium carbonate 1314-61-0,

Tantalum pentoxide 12057-24-8,

Lithium oxide, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(domain switching performance of stoichiometric LiTaO3 prepared

using)

THERE ARE 1 CAPLUS RECORDS THAT CITE OS.CITING REF COUNT: 1

THIS RECORD (1 CITINGS)

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE

FOR THIS RECORD. ALL CITATIONS AVAILABLE

IN THE RE FORMAT

L88 ANSWER 10 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN ACCESSION NUMBER: 1996:338226 HCAPLUS Full-text

DOCUMENT NUMBER: 125:21962

ORIGINAL REFERENCE NO.: 125:4215a,4218a

TITLE: Articles comprising a substrate made of single

crystal and a process for producing the same

INVENTOR(S): Kawaguchi, Tatsuo; Imaeda, Minoru; Fukuda,

Tsuguo

PATENT ASSIGNEE(S): Ngk Insulators, Ltd., Japan

SOURCE: Eur. Pat. Appl., 24 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
  EP 707096	A2	19960417	EP 1995-303946	
				1995 0608
			<	
EP 707096	A3	19970305		
EP 707096	В1	20011017		
R: DE, FR, GB				
US 5650006	A	19970722	US 1995-473964	
				1995
				0607
			<	
JP 09118595	A	19970506	JP 1995-236983	
				1995
				0914
			<	
JP 3725589	В2	20051214		
PRIORITY APPLN. INFO.:			JP 1994-222081	A
				1994
				0916

JP 1995-83939 A
1995
0410
<-JP 1995-213462 A
1995
0822

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

ED Entered STN: 12 Jun 1996

AB Articles (e.g., cptical waveguides) are described which comprise a single crystal substrate supporting a film having a composition of LiNb1-xTaxO3 (O < x ≤ 0.8) formed by LPE. The film has an x-ray rocking curve half width that is not larger than that of the substrate, and the substrate may have the composition LiNb1-zTazO3 (O ≤ z < 0.8; z < x). In forming the films, the substrate is contacted with supercooled liquid phase of a melt consisting mainly of Li2O3, Nb2O5, Ta2O5 and a flux to produce the film. The composition of the liquid phase is within a region encompassed by a straight line K linking a point A (95, 5, 0) and a point B (95, 2, 3), a straight line G linking the point A (95, 5, 0) and a point C (60, 40, 0), a straight line H linking the point C (60, 40, 0) and a point D (60, 0, 40), a straight line J linking the point B (95, 2, 3) and a point E (0, 40, 60) and a curved line I defining a composition whose saturation temperature is not more than 1200° as shown in a triangular diagram of a pseudo-ternary system of LiNbO3-LiTaO3-a melting medium.

IT 1314-61-0, Tantalum oxide

12031-66-2, Lithium tantalate

12057-24-8, Lithium oxide, processes

RL: PEP (Physical, engineering or chemical process); PROC (Process)

(articles comprising lithium niobate tantalate films on a single crystal substrate and their production)

RN 1314-61-0 HCAPLUS

CN Tantalum oxide (Ta2O5) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component	   	Ratio	    +-	Component Registry Number
0	 	3	i	17778-80-2
Ta		1	1	7440-25-7
Li		1		7439-93-2

RN 12057-24-8 HCAPLUS

CN Lithium oxide (Li20) (CA INDEX NAME)

Li\_\_O\_\_Li

```
IC ICM C30B019-02
ICS C30B029-30
```

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 75

IT Waveguides

(  ${\tt optical}, {\tt articles}$  comprising lithium niobate

tantalate films on a single crystal substrate and their production)

IT 12031-63-9, Lithium niobate 115428-32-5, Lithium niobate tantalate (LiNb0.5Ta0.503) 124566-29-6, Lithium niobium tantalum oxide (LiNb0.9Ta0.103) 152126-22-2, Lithium niobium tantalum oxide

```
(LiNb0.35Ta0.6503)
                          169227-36-5, Lithium niobate tantalate
     (LiNb0.84Ta0.1603)
                         169227-37-6, Lithium niobate tantalate
                         169227-38-7, Lithium niobate tantalate
     (LiNb0.59Ta0.4103)
     (LiNb0.78Ta0.2203)
                         169227-39-8, Lithium niobate tantalate
     (LiNb0.32Ta0.6803)
                         169227-40-1, Lithium niobate tantalate
                        169227-41-2, Lithium niobate tantalate
     (LiNb0.64Ta0.3603)
                        169227-42-3, Lithium niobate tantalate
     (LiNb0.88Ta0.1203)
     (LiNb0.83Ta0.1703) 169227-43-4, Lithium niobate tantalate
     (LiNb0.24Ta0.7603) 169227-44-5, Lithium niobate tantalate
     (LiNb0.73Ta0.2703) 169227-45-6, Lithium niobate tantalate
     (LiNb0.85Ta0.1503) 169227-46-7, Lithium niobate tantalate
     (LiNb0.22Ta0.7803) 169227-47-8, Lithium niobate tantalate
     (LiNb0.91Ta0.0903) 169227-48-9, Lithium niobate tantalate
     (\texttt{LiNb0.82Ta0.1803}) \qquad 169227-49-0, \; \texttt{Lithium niobate tantalate}
     (LiNb0.71Ta0.2903) 169227-50-3, Lithium niobate tantalate
     (LiNb0.54Ta0.4603) 177326-82-8, Lithium niobium tantalum
     exide (LiNb0.87Ta0.1303) 177326-83-9, Lithium niobium
     tantalum oxide (LiNb0.8Ta0.203)
                                     177326-84-0,
     Lithium niobium tantalum oxide
     (LiNb0.69Ta0.3103) 177326-85-1, Lithium niobium tantalum
     oxide (LiNb0.56Ta0.4403) 177326-86-2, Lithium niobium
     tantalum oxide (LiNb0.44Ta0.5603) 177326-87-3,
    Lithium niobium tantalum oxide
     (LiNb0.33Ta0.6703) 177326-88-4, Lithium niobium tantalum
     oxide (LiNb0.25Ta0.7503) 177326-89-5, Lithium niobium
    tantalum oxide (LiNb0.76Ta0.2403) 177326-90-8,
     Lithium niobium tantalum oxide
     (LiNb0.75Ta0.2503) 177326-91-9, Lithium niobium tantalum
     oxide (LiNb0.72Ta0.2803)
                              177326-92-0, Lithium niobium
     tantalum oxide (LiNb0.7Ta0.303)
     RL: DEV (Device component use); PEP (Physical, engineering or
     chemical process); PROC (Process); USES (Uses)
        (articles comprising lithium niobate tantalate films on a
        single crystal substrate and their production)
    1313-96-8, Niobium oxide 1314-61-0, Tantalum
     oxide 1314-62-1, Vanadium oxide, processes
     12031-66-2, Lithium tentalate
     12057-24-8, Lithium oxide, processes
     15060-59-0, Lithium vanadate (LiVO3)
     RL: PEP (Physical, engineering or chemical process); PROC
     (Process)
        (articles comprising lithium niobate tantalate films on a
        single crystal substrate and their production)
L88 ANSWER 11 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN
ACCESSION NUMBER: 1996:180857 HCAPLUS Full-text
DOCUMENT NUMBER:
                        124:239810
ORIGINAL REFERENCE NO.: 124:44293a
TITLE:
                        Properties and structures of TeO2 based
                         glasses containing ferroelectric components
AUTHOR(S):
                        Hu, L.; Jiang, Z.
CORPORATE SOURCE:
                        Shanghai Inst. Optics Fine Mechanics, Academia
                         Sinica, Shanghai, 201800, Peop. Rep. China
SOURCE:
                         Physics and Chemistry of Glasses (1996
                        ), 37(1), 19-21
                         CODEN: PCGLA6; ISSN: 0031-9090
PUBLISHER:
                         Society of Glass Technology
DOCUMENT TYPE:
                        Journal
LANGUAGE:
                        English
    Entered STN: 28 Mar 1996
     TeO2 based LiTaO3, KNbO3, PbTiO3 and PbLaTiO3 ferroelec. components containing glasses
     were prepared and examined It was found that PbTiO3 and PbLaTiO3 containing TeO2 based
     glasses have better thermal stability, higher refractive index and larger d. than KNbO3
     and LiTaO3 containing TeO2 based glasses. From IR and Raman spectra results it is
     deduced that PbTiO3 and PbLaTiO3 containing TeO2 based glasses consist of sym. TeO4
     trigonal bipyramids and deformed TeO4 groups and that LiTaO3 and KNbO3 containing TeO2
     based glasses consisting of TeO4 trigonal bipyramids and TeO3 trigonal pyramids. It is
```

тт

```
confirmed that the structural transition from TeO4 to TeO3 deteriorates the thermal
     stability in LiTaO3 and KNbO33 containing TeO2 based glasses.
    1314-61-0, Tantalum oxide
    12031-66-2, Lithium tantalate (
    LiTaO3) 12057-24-8, Lithium
    oxide, properties
    RL: PRP (Properties); TEM (Technical or engineered material use);
    USES (Uses)
       (glass; properties and structures of TeO2 based glasses containing
       ferroelec. components)
    1314-61-0 HCAPLUS
RN
CN
    Tantalum oxide (Ta2O5) (CA INDEX NAME)
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
   12031-66-2 HCAPLUS
   Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)
CN
                   Ratio |
 Component |
                                      Component
                                | Registry Number
            __________
                                    17778-80-2
               3
\circ
            Ta
                     1
                                - 1
                                        7440-25-7
             - 1
                               7439-93-2
                     1
Li
             12057-24-8 HCAPLUS
RN
CN Lithium oxide (Li20) (CA INDEX NAME)
Li__O__Li
CC
   57-1 (Ceramics)
TT
    Density
    Ferroelectric substances
    Glass temperature and transition
    Infrared spectra
    Raman spectra
    Refractive index and Optical refraction
       (properties and structures of TeO2 based glasses containing
       ferroelec. components)
ТТ
    1312-81-8, Lanthanum sesquioxide 1313-96-8, Niobium oxide
    1314-61-0, Tantalum oxide 1317-36-8,
    Lead monoxide, properties 7446-07-3, Tellurium oxide (TeO2)
    12030-85-2, Potassium niobate (KNbO3) 12031-66-2,
    Lithium tantalate (LiTaO3)
    12057-24-8, Lithium oxide, properties
    12060-00-3, Lead titanium oxide (PbTiO3) 12136-45-7, Potassium
    oxide, properties 13463-67-7, Titania, properties 114952-68-0,
    Lanthanum lead titanium oxide (LaPbTiO3)
    RL: PRP (Properties); TEM (Technical or engineered material use);
    USES (Uses)
       (glass; properties and structures of TeO2 based glasses containing
       ferroelec. components)
OS.CITING REF COUNT: 15
                             THERE ARE 15 CAPLUS RECORDS THAT CITE
                             THIS RECORD (16 CITINGS)
L88 ANSWER 12 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN
ACCESSION NUMBER: 1980:576211 HCAPLUS Full-text
DOCUMENT NUMBER:
                      93:176211
ORIGINAL REFERENCE NO.: 93:27933a,27936a
TITLE:
                       Refractive indexes and surface morphology of
                       LPE grown lithium (niobium, tantalum
                       ) oxide (Li(Nb, Ta)O3) films on
                       lithium tantalate(V)
```

substrates

AUTHOR(S): Kondo, Susumu; Miyazawa, Shintaro; Sugii,

Kiyomassa; Iwasaki, Hiroshi

CORPORATE SOURCE: Musashino Electr. Commun. Lab., Nippon Telegr.

and Teleph. Public Corp., Musashino, 180,

Japan

SOURCE: Journal of Crystal Growth (1980),

50(3), 605-11

CODEN: JCRGAE; ISSN: 0022-0248

DOCUMENT TYPE: Journal LANGUAGE: English ED Entered STN: 12 May 1984

AB Li(Nb,Ta)O3 solid-solution films were grown on LiTaO3 substrates from a Li2O-V2O5 flux using an LPE technique. For obtaining the films suitable for optical waveguide applications, as-grown surface morphol., lattice parameter mismatches and refractive index differences between the film and the substrate were studied with regard to the solution composition Li2O content and Nb2O5/Ta2O5 ratio in the starting solution (Li2O-(Nb2O5, Ta2O5)-V2O5 affect the refractive indexes of the grown films. It was clarified that the ordinary and extraordinary refractive index differences, Δno and Δne, can be controlled independently in the range of 0-5 + 10-3 by varying the solution composition

IT 1314-61-00, solid solns. with lithium
omide and niobium oxide 12031-66-20, solid
solns. with lithium niobate 12057-24-80, solid solns.
with niobium oxide and tantalum omide
RL: PRP (Properties)

(refractive index of liquid phase epitaxially grown)

RN 1314-61-0 HCAPLUS

CN Tantalum oxide (Ta2O5) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component		Ratio	   	Component Registry Number
0	+ 	3	+- 	17778-80-2
Ta	j	1	i	7440-25-7
Li	1	1	1	7439-93-2

RN 12057-24-8 HCAPLUS

CN Lithium oxide (Li20) (CA INDEX NAME)

Li\_O\_Li

CC 73-2 (Spectra by Absorption, Emission, Reflection, or Magnetic Resonance, and Other Optical Properties) Section cross-reference(s): 75

ST refraction lithium niobium tantalum oxide

IT Refractive index and Optical refraction (of lithium niobium tantalum cxide films)

IT Crystal structure

Surface structure

(of lithium niobium tantalum oxide films grown by liquid phase epitaxy on lithium tantalate substrates)

IT 1313-96-8D, solid solns. with lithium oxide and tantalum oxide 1314-61-0D, solid solns. with lithium oxide and niobium oxide 12031-63-9D, solid solns. with lithium tantalate 12031-66-2D, solid solns. with

lithium niobate 12057-24-8D, solid solns. with niobium oxide and tantalum oxide RL: PRP (Properties)

(refractive index of liquid phase epitaxially grown)

L88 ANSWER 13 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN ACCESSION NUMBER: 1973:519157 HCAPLUS Full-text

DOCUMENT NUMBER: 79:119157

ORIGINAL REFERENCE NO.: 79:19329a,19332a

TITLE: Stoichiometry and optical quality of

lithium tantalate(V) single

crystals

AUTHOR(S): Miyazawa, Shintaro; Iwasaki, Hiroshi

CORPORATE SOURCE: Musashino Electr. Commun. Lab., Nippon Telegr.

and Teleph. Public Corp., Musashino, Japan Review of the Electrical Communications

Laboratories (1973), 21(5-6), 374-83

CODEN: RELTAN; ISSN: 0029-067X

DOCUMENT TYPE: Journal LANGUAGE: English ED Entered STN: 12 May 1984

The phase diagram of the %:20-%\*205 system for starting melt compns. Cl = 49.0-54.25 mole % Ta205 was determined by measuring the ferroelec. Curie temps. (TC) of the crystals cooled from the melts. The composition of the congruently melting solid (with TC = 618°) was Cl = 51.25 mole % Ta205. Single crystals grown from the melt with Cl = 51.25 mole % Ta205 by the Czochralski method with the temperature controlled to ±0.25° during the crystal pulling with optically homogeneous ( bixefringence variation of <10-5 throughout the crystal). Such crystals were homogeneous in composition and TC value throughout the boule. For crystals with the congruent-melting composition, the ratio of the min. to the maximum transmitted light intensities (λ = 6328 Å from a He-Ne laser) increased from 0.25 to 3% as the light-beam diameter was increased from 1 to 6 mm, resp.

IT 12031-66-2

SOURCE:

RL: PEP (Physical, engineering or chemical process); PROC (Process)

(crystal growth of, elec. and optical properties in relation to)

RN 12031-66-2 HCAPLUS

CN Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)

Component	   	Ratio		Component Registry Number
	==+==		===+=	
0	- 1	3	1	17778-80-2
Ta	1	1	1	7440-25-7
Li	1	1	1	7439-93-2

IT 1314-61-0

RL: PRP (Properties)

(system, lithium oxide-)

RN 1314-61-0 HCAPLUS

CN Tantalum oxide (Ta2O5) (CA INDEX NAME)

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

IT 12057-24-8

RL: PRP (Properties)

(system, tantalum oxide-)

RN 12057-24-8 HCAPLUS

CN Lithium oxide (Li2O) (CA INDEX NAME)

Li\_\_O\_\_Li

```
CC
     70-1 (Crystallization and Crystal Structure)
     Section cross-reference(s): 68, 73, 71
     lithium tantalum oxide system; Curie temp
     lithium tantalate; single crystal
     lithium tantalate; birefringence
     lithium tantalate
    Crystal growth
TТ
     Curie point, ferroelectric
        (of lithium tantalum oxide)
     Birefringence
        (of lithium tantalum oxide single crystals)
IT
     12031-66-2
     RL: PEP (Physical, engineering or chemical process); PROC
        (crystal growth of, elec. and optical properties in
        relation to)
     1314-61-0
ΙT
     RL: PRP (Properties)
        (system, lithium oxide-)
TΤ
     12057-24-8
     RL: PRP (Properties)
        (system, tantalum oxide-)
OS.CITING REF COUNT: 3
                              THERE ARE 3 CAPLUS RECORDS THAT CITE
                               THIS RECORD (3 CITINGS)
L88 ANSWER 14 OF 16 HCAPLUS COPYRIGHT 2010 ACS on STN
ACCESSION NUMBER:
                         1973:76711 HCAPLUS Full-text
DOCUMENT NUMBER:
                         78:76711
ORIGINAL REFERENCE NO.: 78:12169a,12172a
TITLE:
                         Stoichiometry and optical quality of
                         lithium tantalate single
                         crystals
AUTHOR(S):
                         Miyazawa, Shintaro; Iwasaki, Hiroshi
CORPORATE SOURCE:
                         Musashino Electr. Commun. Lab., Nippon Telegr.
                         Teleph. Public Corp., Musashino, Japan
SOURCE:
                         Kenkyu Jitsuyoka Hokoku - Denki Tsushin
                         Kenkyusho (1972), 21(9), 1739-51
                         CODEN: DTKKAA; ISSN: 0415-3200
DOCUMENT TYPE:
                         Journal
                         Japanese
ED
     Entered STN: 12 May 1984
     The phase diagram of the system 1.120-7 \approx 205 was determined from the composition
AB
     dependence of the Curie temperature of LiTaO3. The congruent composition was found to
     be 48.75/51.25 in Li/Ta molar ratio, at which optically homogeneous single crystals of
     LiTaO3 can be grown from the melt. The extinction ratio of these crystals was 1.5% for
     the 6328 \hbox{\normalfont\AA} light beam of about 4 mm diameter. On the other hand, change in
     birefringence of a crystal grown from the stoichiometric melt (Li/Ta = 50/50) was
     measured and agrees with calculated values. In order to grow high quality crystals from
     the stoichiometric melt, it is required to control the temperature within \pm 0.3^{\circ}.
     12031-66-2
ΤТ
     RL: PEP (Physical, engineering or chemical process); PROC
     (Process)
        (crystal growth of, optical quality and stoichiometry
        in relation to)
     12031-66-2 HCAPLUS
RN
CN
     Lithium tantalum oxide (LiTaO3) (CA INDEX NAME)
```

Component	- 1	Ratio		Component
			- 1	Registry Number
	==+==		===+==	
0	1	3	1	17778-80-2
Ta	1	1	1	7440-25-7
Li		1	- 1	7439-93-2

TT 1314-61-0

RL: PRP (Properties)

(system, lithium oxide-)

```
BN
    1314-61-0 HCAPLUS
CN Tantalum oxide (Ta2O5) (CA INDEX NAME)
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
ΙT
   12057-24-8
    RL: PRP (Properties)
     (system, tantalum oxide-)
RN
   12057-24-8 HCAPLUS
CN Lithium oxide (Li20) (CA INDEX NAME)
Li__O__Li
    70-1 (Crystallization and Crystal Structure)
ST
    growth lithium tantalum oxide
IT Optical property
      (of lithium tantalate single crystals)
ΤТ
   Crystal growth
       (of lithium tantalate, optical
       quality and stoichiometry in relation to)
ΙT
    12031-66-2
    RL: PEP (Physical, engineering or chemical process); PROC
    (Process)
       (crystal growth of, optical quality and stoichiometry
       in relation to)
ΙT
    1314-61-0
    RL: PRP (Properties)
       (system, lithium oxide-)
IT
    12057-24-8
    RL: PRP (Properties)
      (system, tantalum oxide-)
L88 ANSWER 15 OF 16 WPIX COPYRIGHT 2010
                                         THOMSON REUTERS on STN
ACCESSION NUMBER: 1984-103832 [198417]
                                       WPIX
DOC. NO. CPI:
                   C1984-044016 [199321]
TITLE:
                   Lithium tantalate single
                   crystal vertical drawing - from melt of tantalum
                    pent:oxide and lithium oxide
                   in e.g. a crucible of iridium
                E31; L03
DERWENT CLASS:
                  ENOKIDA K; TSUNODA M
INVENTOR:
PATENT ASSIGNEE: (TOKE-C) TOKYO SHIBAURA DENKI KK
COUNTRY COUNT:
PATENT INFO ABBR.:
     PATENT NO KIND DATE WEEK LA PG
                                                     MAIN IPC
     JP 59045999 A 19840315 (198417) * JA 2[0]
          <--
APPLICATION DETAILS:
     PATENT NO
                 KTND
                                      APPLICATION DATE
     ______
     JP 59045999 A
                                       JP 1982-153842
          19820906
PRIORITY APPLN. INFO: JP 1982-153842
                                      19820906
ED 20050420
AN 1984-103832 [198417] WPIX
AB JP 59045999 A UPAB: 20050420
```

Tantalum pentoxide and lithium oxide in a prescribed molar ratio are charged directly into a crucible of platinum-platinum rhodium or iridium. After melting them, a single crystal of lithium tantalate is vertically drawn.

Inclusion of impurities can be decreased. Sintering and grinding steps are not required. Defects during the growth of the single crystal can be obviated. The expensive raw materials can be efficiently used. The cost of the crystal substrate of LiTaO3 can be lowered and quality surface wave resilient elements provided.

L88 ANSWER 16 OF 16 WPIX COPYRIGHT 2010 THOMSON REUTERS on STN

ACCESSION NUMBER: 1977-47926Y [197727] WPIX TITLE: Lithium tantalate light

modulating element production - using melt of

tantalum oxide and

lithium oxide to give single lithium tantalate crystal

DERWENT CLASS: E31; L03; P81; V07

TSUYA H INVENTOR:

PATENT ASSIGNEE: (NIDE-C) NIPPON ELECTRIC CO

COUNTRY COUNT:

PATENT INFO ABBR.:

PATENT NO KIND DATE WEEK LA PG MAIN IPC JP 52063743 A 19770526 (197727)\* JA

<--

JP 58048519 B 19831028 (198347) JA

<--

APPLICATION DETAILS:

APPLICATION DATE PATENT NO KIND JP 52063743 A JP 1975-140516

19751121

PRIORITY APPLN. INFO: JP 1975-140516 19751121

20050417

AN 1977-47926Y [197727] WPIX

AΒ JP 52063743 A UPAB: 20050417

> The LiTaO2 light modulation element is produced from single crystal of lithium tantalate which is bred from the melt of tantalum oxide and lithium oxide having mol. ratio of Li/Ta of 1.15-1.25 by polarisation under 60-40 V/cm. electric field and heat treatment at 400-500 degrees C in oxygen atmos.

DATA NOT AVAILABLE FOR THIS ACCESSION NUMBER

=> d 188 15-16 full

L88 ANSWER 15 OF 16 WPIX COPYRIGHT 2010 THOMSON REUTERS on STN

AN 1984-103832 [198417] WPIX Full-text

DNC C1984-044016 [199321]

TI Lithium tantalate single crystal vertical

drawing - from melt of tantalum pent:oxide and lithium

oxide in e.g. a crucible of iridium

DC E31; L03

IMENOKIDA K; TSUNODA M

(TOKE-C) TOKYO SHIBAURA DENKI KK

CYC 1

JP 59045999 A 19840315 (198417) \* JA 2[0] PΙ

<--

ADT JP 59045999 A JP 1982-153842 19820906

PRAI JP 1982-153842 19820906

```
IPCR C30B0015-00 [I,A]; C30B0015-00 [I,C]; C30B0029-10 [I,C];
    C30B0029-30 [I.A]
EPC C30B0015-00+29/30
     JP 59045999 A UPAB: 20050420
       Tantalum pentoxide and lithium oxide in a prescribed molar ratio are charged directly
     into a crucible of platinum-platinum rhodium or iridium. After melting them, a single
     crystal of lithium tantalate is vertically drawn.
     Inclusion of impurities can be decreased. Sintering and grinding steps are not
     required. Defects during the growth of the single crystal can be obviated. The
     expensive raw materials can be efficiently used. The cost of the crystal substrate of
     LiTaO3 can be lowered and quality surface wave resilient elements provided.
FS
MC
    CPI: E35-N; L02-A09
L88 ANSWER 16 OF 16 WPIX COPYRIGHT 2010
                                               THOMSON REUTERS on STN
    1977-47926Y [197727] WPIX Full-text
AN
    Lithium tantalate light modulating element
ΤI
    production - using melt of tantalum oxide and
    lithium oxide to give single lithium
    tantalate crystal
    E31; L03; P81; V07
DC
TN
   TSUYA H
PA
    (NIDE-C) NIPPON ELECTRIC CO
CYC 1
PΙ
   JP 52063743
                   A 19770526 (197727)* JA
    JP 58048519
                   B 19831028 (198347) JA
    <--
ADT JP 52063743 A JP 1975-140516 19751121
PRAI 30 1975-140516
                         19751121
IPCR C01G0035-00 [I,A]; C01G0035-00 [I,C]; C09K0003-00 [I,A];
    C09K0003-00 [I,C]; C30B0029-10 [I,C]; C30B0029-30 [I,A];
    C30B0033-00 [I,A]; C30B0033-00 [I,C]; C30B0033-04 [I,A];
    G02F0001-01 [I,C]; G02F0001-03 [I,A]; G02F0001-05 [I,A];
    H01L0041-24 [I,A]; H01L0041-24 [I,C]
AB
    JP 52063743 A UPAB: 20050417
     The LiTaO2 light modulation element is produced from single crystal of lithium
     tantalate which is bred from the melt of tantalum oxide and lithium oxide having mol.
     ratio of Li/Ta of 1.15-1.25 by polarisation under 60-40 V/cm. electric field and heat
     treatment at 400-500 degrees C in oxygen atmos.
FS
    CPI; GMPI; EPI
MC
    CPI: E35-N; L02-G07; L03-D04; L03-G02
```

#### FULL SEARCH HISTORY

=> d his nofile

L5

(FILE 'HOME' ENTERED AT 12:55:42 ON 07 MAY 2010)

FILE 'HCAPLUS' ENTERED AT 12:55:49 ON 07 MAY 2010

E US20090108232/PN

L11 SEA SPE=ON ABB=ON PLU=ON US20090108232/PN

D ALL SEL RN

FILE 'REGISTRY' ENTERED AT 12:58:40 ON 07 MAY 2010

3 SEA SPE=ON ABB=ON PLU=ON (12031-66-2/BI OR 12057-24-L2

8/BI OR 1314-61-0/BI)

D SCA

FILE 'REGISTRY' ENTERED AT 13:03:10 ON 07 MAY 2010

FILE 'HCAPLUS' ENTERED AT 13:03:24 ON 07 MAY 2010

FILE 'REGISTRY' ENTERED AT 13:03:34 ON 07 MAY 2010

1 SEA SPE=ON ABB=ON PLU=ON 12031-66-2/RN L3

D SCA

L41 SEA SPE=ON ABB=ON PLU=ON 1314-61-0/RN

1 SEA SPE=ON ABB=ON PLU=ON 12057-24-8/RN

FILE 'HCAPLUS' ENTERED AT 13:04:32 ON 07 MAY 2010

FILE 'REGISTRY' ENTERED AT 13:04:51 ON 07 MAY 2010

D L3 CN

D L4 CN

D L5 CN

FILE 'HCAPLUS' ENTERED AT 13:05:36 ON 07 MAY 2010 1.6

4831 SEA SPE=ON ABB=ON PLU=ON L3

FILE 'REGISTRY' ENTERED AT 13:05:52 ON 07 MAY 2010

SET SMARTSELECT ON

SEL PLU=ON L3 1- NAME : 5 TERMS L7

SET SMARTSELECT OFF

FILE 'HCAPLUS' ENTERED AT 13:05:52 ON 07 MAY 2010

4918 SEA SPE=ON ABB=ON PLU=ON L7 L8

5233 SEA SPE=ON ABB=ON PLU=ON L6 OR L8 L9

19677 SEA SPE=ON ABB=ON PLU=ON L4 L10

FILE 'REGISTRY' ENTERED AT 13:06:15 ON 07 MAY 2010

SET SMARTSELECT ON

L11 SEL PLU=ON L4 1- NAME : 12 TERMS

SET SMARTSELECT OFF

FILE 'HCAPLUS' ENTERED AT 13:06:16 ON 07 MAY 2010

30017 SEA SPE=ON ABB=ON PLU=ON L11 L12

31488 SEA SPE=ON ABB=ON PLU=ON L1 18247 SEA SPE=ON ABB=ON PLU=ON L5 L13 PLU=ON L10 OR L12

L14

FILE 'REGISTRY' ENTERED AT 13:06:58 ON 07 MAY 2010

SET SMARTSELECT ON

L15 SEL PLU=ON L5 1- NAME : 5 TERMS

SET SMARTSELECT OFF

FILE 'HCAPLUS' ENTERED AT 13:06:58 ON 07 MAY 2010

L16 28098 SEA SPE=ON ABB=ON PLU=ON L15

L17 15108 SEA SPE=ON ABB=ON PLU=ON L14 AND L16

			10/3	73,742-3	JUIJU-LIC SLAKCII
L18	12990	SEA SPE=ON	ABB=ON	PLU=ON	L16 NOT L17
L19	4516	SEA SPE=ON	ABB=ON	PLU=ON	L6 AND L8
L20	18206	SEA SPE=ON	ABB=ON	PLU=ON	L10 AND L12
L21		SEA SPE=ON	ABB=ON	PLU=ON	L17 AND L19 AND L20
L22					(MOLAR OR MOLE)(4A)RATIO
L23	3	D SCA	ABB=ON	PLU=ON	L21 AND L22
		D 1-3 KWIC			
L24	1	SEA SPE=ON D SCA D ABS	ABB=ON	PLU=ON	L1 AND L23
L25	3		ABB=ON	PLU=ON	(LIO2)(2W)(TA2O5)
L26	2	D KWIC D 2	ABB=ON	PLU=ON	0.975(3W)0.982
L27	18	D KWIC 2 SEA SPE=ON D KWIC	ABB=ON	PLU=ON	GTOREQ(4A)0.975
L28	10		ABB=ON	PLU=ON	LTOREQ(4A)0.982
L29					L27 AND L28
L30					((LITHIUM OR DILITHIUM)(A)(
		OXIDE OR DI TANTALUM(A) D KWIC	OXIDE) O	R LIO2 O	R O2LI OR LI2O OR OLI2)(5W)(
- 0.	1.0	D 5 KWIC			- 00 00
L31	. 10	SEA SPE=ON D KWIC D 5 KWIC	ABB=ON	PLU=ON	L30 AND L22
L32	1		ABB=ON	PI.II=ON	L31 AND (L26 OR L27 OR
102		L28) D KWIC	ADD-ON	1 110-011	HIST AND VIEW ON HE / ON
L33	1		ABB=ON	PLU=ON	L32 AND L21
L34					L21 AND L22
L35					L21 AND ((L22 OR L23 OR
					L28 OR L29 OR L30 OR L31
		OR L32 OR L D KWIC D 3 KWIC D SCA L1 E BIREFRING	.33 OR L3	4))	
		E E3+ALL			
L36 L37		SEA SPE=ON SEA SPE=ON D KWIC D 2 KWIC	ABB=ON ABB=ON	PLU=ON PLU=ON	BIREFRINGENCE+MAX/CT L36 AND L21
L38	37641	SEA SPE=ON	ABB=ON	PLH=ON	BIREFRING?
L39		SEA SPE=ON D KWIC	ABB=ON	PLU=ON	
L40	19976	SEA SPE=ON D KWIC	ABB=ON	PLU=ON	0.0005
L41	19976	SEA SPE=ON	ABB=ON	PLU=ON	0.0005
L42	850	SEA SPE=ON D KWIC	ABB=ON	PLU=ON	L40 (4A) L41
L43		SEA SPE=ON			L40(L)(L36 OR L38)
L44		SEA SPE=ON			L41(L)(L36 OR L38)
L45	45	SEA SPE=ON D KWIC	ABB=ON	PLU=ON	L43 AND L44
L46	1	D 30 KWIC SEA SPE=ON D KWIC	ABB=ON	PLU=ON	L42(L)(L36 OR L38)
L47		QUE SPE=ON	ABB=ON	PLU=ON	LENS? OR OPTIC? OR OPTO?
L48		SEA SPE=ON D KWIC			L21 AND L47
L49					L48 AND (L36 OR L38)
L50	1	SEA SPE=ON L42 OR L43			L49 AND ((L40 OR L41 OR L46))

	D KWIC			
L51		ABB=ON	PI.II=ON	1.48 AND 1.22
шэт	D KWIC	1100-011	1 10-011	110 1110 122
	D KWIC 2			
	D KWIC 3			
L52		ABB=ON	PLU=ON	L48 AND (L22 OR L36 OR
	L38)			·
	D SCA			
L53		ABB=ON	PLU=ON	L52 AND ((L23 OR L24 OR
	L25 OR L26 C			
L54	4 SEA SPE=ON	ABB=ON	PLU=ON	L52 AND L30
L55	4 SEA SPE=ON	ABB=ON	PLU=ON	L53 OR L54
L56				L52 AND ((L40 OR L41 OR
	L42 OR L43 C	R L44 OF	R L45 OR	L46))
	D KWIC			
L57	17 SEA SPE=ON	ABB=ON	PLU=ON	(L48 OR L49 OR L50 OR L51
	OR L52 OR L5	3 OR L54	1 OR L55	OR L56)
L58	1 SEA SPE=ON	ABB=ON	PLU=ON	L57 AND L1
	D KWIC			
L59				PY=<2005 NOT P/DT
L60				(PY=<2005 OR PRY=<2005 OR
				IEW/DT) AND P/DT
L61	14 SEA SPE=ON	ABB=ON	PLU=ON	L57 AND (L59 OR L60)
				AT 13:49:04 ON 07 MAY 2010
L62				
L63		ABB=ON	PLU=ON	L 4
	D KWIC			_
L64	8 SEA SPE=ON			
L65	0 SEA SPE=ON			
L66	145 SEA SPE=ON			
L67				
L68				
L69	U SEA SPE=ON	ABB=ON	PLU=ON	L66 AND L67 AND L68
	ETTE IDTSCARS DASCAI	COMECCI	TARTO	WPIX' ENTERED AT 13:51:44
	ON 07 MAY 2010	COMP SCI,	JAPIO,	WPIA ENIERED AI 13.31.44
L70		ARR-ON	DI II-ON	17
	12031 SEA SPE=ON			
L72				
L73				L70 AND L71 AND L72
1,0	D KWIC	1100 011	110 01	HIVE HIVE HIVE HIVE
L74		ABB=ON	PLU=ON	L73 AND L22
	D KWIC			
L75				
		ABB=ON	PLU=ON	L73 AND L38
L76	D KWIC	ABB=ON	PLU=ON	L73 AND L38
L77	1 SEA SPE=ON D KWIC	ABB=ON	PLU=ON	
L77 L78	1 SEA SPE=ON D KWIC 5 SEA SPE=ON	ABB=ON ABB=ON	PLU=ON	L74 AND L75
	1 SEA SPE=ON D KWIC 5 SEA SPE=ON	ABB=ON ABB=ON ABB=ON	PLU=ON	L74 AND L75 (L74 OR L75 OR L76)
	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON	ABB=ON ABB=ON ABB=ON	PLU=ON	L74 AND L75 (L74 OR L75 OR L76)
	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30	ABB=ON ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON	L74 AND L75 (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR
L78	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30	ABB=ON ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON	L74 AND L75 (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR
L78	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30 2 SEA SPE=ON D SCA 1 SEA SPE=ON	ABB=ON ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON	L74 AND L75 (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR
L78 L79	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30 2 SEA SPE=ON D SCA	ABB=ON ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON	L74 AND L75  (L74 OR L75 OR L76)  L77 AND ((L25 OR L26 OR  L77 AND L30
L78 L79	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30 2 SEA SPE=ON D SCA 1 SEA SPE=ON D KWIC	ABB=ON ABB=ON ABB=ON ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON PLU=ON	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)
L78 L79 L80	1 SEA SPE=ON	ABB=ON ABB=ON ABB=ON ABB=ON ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)
L78 L79	1 SEA SPE=ON	ABB=ON ABB=ON ABB=ON ABB=ON ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)
L78 L79 L80	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30 2 SEA SPE=ON D SCA 1 SEA SPE=ON D KWIC  FILE 'HCAPLUS' ENTERED 18 SEA SPE=ON D KWIC	ABB=ON ABB=ON ABB=ON ABB=ON ABB=ON AT 13:59 ABB=ON	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)  07 MAY 2010 0.0005(3A)(PLUS OR MINUS)
L78 L79 L80	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30 2 SEA SPE=ON D SCA 1 SEA SPE=ON D KWIC  FILE 'HCAPLUS' ENTERED 18 SEA SPE=ON D KWIC	ABB=ON ABB=ON ABB=ON ABB=ON ABB=ON AT 13:59 ABB=ON	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)  07 MAY 2010 0.0005(3A)(PLUS OR MINUS)
L78 L79 L80	1 SEA SPE=ON	ABB=ON ABB=ON ABB=ON ABB=ON AT 13:59 ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)  07 MAY 2010 0.0005(3A)(PLUS OR MINUS)  L81 AND L39
L78 L79 L80	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30 2 SEA SPE=ON D SCA 1 SEA SPE=ON D KWIC  FILE 'HCAPLUS' ENTERED 18 SEA SPE=ON D KWIC 0 SEA SPE=ON FILE 'DISSABS, PASCAL,	ABB=ON ABB=ON ABB=ON ABB=ON AT 13:59 ABB=ON ABB=ON	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)  07 MAY 2010 0.0005(3A)(PLUS OR MINUS)
L78 L79 L80 L81 L82	1 SEA SPE=ON D KWIC 5 SEA SPE=ON 1 SEA SPE=ON 1 SEA SPE=ON L27 OR L28)) D QUE L30 2 SEA SPE=ON D SCA 1 SEA SPE=ON D KWIC  FILE 'HCAPLUS' ENTERED 18 SEA SPE=ON D KWIC 0 SEA SPE=ON FILE 'DISSABS, PASCAL, ON 07 MAY 2010	ABB=ON ABB=ON ABB=ON ABB=ON AT 13:59 ABB=ON ABB=ON CONFSCI,	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON JAPIO,	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)  07 MAY 2010 0.0005(3A)(PLUS OR MINUS) L81 AND L39  WPIX' ENTERED AT 14:00:40
L78 L79 L80	1 SEA SPE=ON	ABB=ON ABB=ON ABB=ON ABB=ON AT 13:59 ABB=ON ABB=ON CONFSCI,	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON JAPIO,	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)  07 MAY 2010 0.0005(3A)(PLUS OR MINUS) L81 AND L39  WPIX' ENTERED AT 14:00:40
L78 L79 L80 L81 L82	1 SEA SPE=ON	ABB=ON ABB=ON ABB=ON ABB=ON ABB=ON AT 13:59 ABB=ON CONFSCI, ABB=ON	PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON PLU=ON JAPIO, PLU=ON	L74 AND L75  (L74 OR L75 OR L76) L77 AND ((L25 OR L26 OR  L77 AND L30  L77 AND (L40 OR L41)  07 MAY 2010 0.0005(3A)(PLUS OR MINUS) L81 AND L39  WPIX' ENTERED AT 14:00:40

		SAV	TEMP L8	4 PEE942	MULT/A	
L85	0	SEA	SPE=ON	ABB=ON	PLU=ON	L84 AND L59
L86	3	SEA	SPE=ON	ABB=ON	PLU=ON	L84 AND L60
L87	3	SEA	SPE=ON	ABB=ON	PLU=ON	L85 OR L86
		SAV	TEMP L8	7 PEE942	MULT/A	

FILE 'HCAPLUS' ENTERED AT 14:08:48 ON 07 MAY 2010 SAV TEMP L61 PEE942HCP/A

FILE 'STNGUIDE' ENTERED AT 14:09:18 ON 07 MAY 2010

D QUE L61
D QUE L69
D QUE L87

OR L83

FILE 'HCAPLUS, WPIX' ENTERED AT 14:12:26 ON 07 MAY 2010
L88

16 DUP REM L61 L69 L87 (1 DUPLICATE REMOVED)
ANSWERS '1-14' FROM FILE HCAPLUS
ANSWERS '15-16' FROM FILE WPIX
D L88 1-16 IBIB ED ABS HITSTR HITIND

D L88 15-16 FULL